

Reference - Valency Chart

POSITIVE IONS [Basic Radicals]

MONOVALENT VALENCY 1		DIVALENT VALENCY 2		TRIVALENT VALENCY 3	
• POTASSIUM	K^{1+}	• CALCIUM	Ca^{2+}	• ALUMINIUM	Al^{3+}
• SODIUM	Na^{1+}	• MAGNESIUM	Mg^{2+}	• CHROMIUM	Cr^{3+}
• LITHIUM	Li^{1+}	• ZINC	Zn^{2+}	• FERRIC	Iron [III] Fe^{3+}
• HYDROGEN	H^{1+}	• BARIUM	Ba^{2+}	• AURIC	Gold [III] Au^{3+}
• AMMONIUM	NH_4^{1+}	• NICKEL	Ni^{2+}		
• CUPROUS	Copper [I] Cu^{1+}	• CUPRIC	Copper [II] Cu^{2+}		
• ARGENTOUS	Silver [I] Ag^{1+}	• ARGENTIC	Silver [II] Ag^{2+}		
• MERCUROUS	Mercury [I] Hg^{1+}	• MERCURIC	Mercury [II] Hg^{2+}		
• AUROUS	Gold [I] Au^{1+}	• FERROUS	Iron [II] Fe^{2+}		
		• PLUMBOUS	Lead [II] Pb^{2+}	• PLUMBIC	Lead [IV] Pb^{4+}
		• STANNOUS	Tin [II] Sn^{2+}	• STANNIC	Tin [IV] Sn^{4+}
		• PLATINOUS	Platinum [II] Pt^{2+}	• PLATINIC	Platinum [IV] Pt^{4+}
		• MANGANOUS	Manganese [II] Mn^{2+}	• MANGANIC	Manganese [IV] Mn^{4+}

NEGATIVE IONS [Acidic Radicals]

[illegible]